

Analysis of Macro and Micronutrients in villages of Vadagam Taluka, Banaskantha (Gujarat) India.

Jaimini A. Patel¹, Kinjal B. Chaudhary¹, Pradhuman A. Parmar¹ & Upendra R. Patel^{1*}

^{1*} Department of Chemistry, M.N. College Visnagar, Gujarat, India.

Abstract:

Present study of Macro and Micronutrients of soil analysis of Vadagam Taluka, Dist. Banaskantha, Gujarat is based on various parameters like pH, Electrical conductivity, available potash, phosphorous, organic carbon, available sulphur, calcium, magnesium, manganese, zinc, copper and iron. Five representative locations were selected for the study and 20 samples from each location and direction of area were collected. This study leads us to the conclusion of Nutrients status of the region and based on analysis we can recommend to the concern farmer for fertilizers. Low, medium & high range of above parameter also calculated from analysis data.

Keywords: Soil, Electrical conductivity, Fertilizers, Available Sulphur.

Introduction:

The particle shape of sediments, soils, and rocks, moisture contents have influences on important transport properties, such as electrical conductivity [1], dielectric permittivity [2], soil resistivity, thermal conductivity, and hydraulic conductivity. Soil electrical resistivity depends on soil water content as well as dry density of data fields [3]. Soil resistivity also depends on soil texture (especially content of clay), soil type and water holding capacity and the amount of dissolved ions in pore water [4-5]. More dry density reduces water content of soil which increases the resistivity of surface soil in geotechnical field.

This is especially true in water limited areas, of which there are many distributed over the globe. Moreover in Europe, a direct link is observed between soil water status, gross primary productivity of vegetation and soil respiration [6].

Sustainable agriculture aims at long term maintenance of natural resources and agricultural productivity with minimal adverse impact on the environment. It emphasizes optimal crop production with minimal external inputs, reducing dependence on commercial inputs (fertilizers and pesticides) and substituting them with internal resources. Various researchers have studied on physico – chemical characterization of farmland soil [7 - 9].

Plant Nutrients

Plants absorb a large number of elements, all of them are essential or not essential for the growth of plant. Some elements which are required by plant for their normal growth, development, metabolism and to complete their life cycle are called the essential elements. Some of these are required in large amounts and some in traces. Nutrients are classified as Primary (Macro), Secondary and micro, and are further classified as follow:

Major nutrients required for plant growth - Required in large amounts.

Class 1: Carbon, hydrogen and oxygen (C, H, O)

Class 2: Nitrogen, potassium and phosphorus

Secondary Nutrients: – Required in medium amounts.

Magnesium, Calcium and sulphur

Micro nutrients: – Required in trace amounts.

Iron, boron, zinc, molybdenum, manganese, copper and chlorine

Soil testing can be divided into four steps (1) sampling (2) analysis (3) interpretation and (4) recommendations. One of the most important aspects of soil testing is that of obtaining a representative sample of the area.

CHEMICAL & EQUIPMENTS

Potassium chloride, Buffer tablate, Sulphuric acid, Potassium dichromate, Sodium bicarbonate, activated charcoal (phosphorous free), Ammonium molybdate, Stannous chloride, Ammonium acetate, Calcium chloride, Glacial acetic acid, Barium chloride, Gum acacia, Sodium diethyl dithiocarbamate, Sodium hydroxide, Muroxide, Ethylene di amine tetraacetate, Ammonia buffer, Diethylenetriamine pentaacetic acid, Eriochrome black-T, were procured from s.d. fine chem Ltd. All chemicals are of analytical grade reagent.

pH was measured on pH meter (systronics Model No-335), Conductivity was measured on conductivity meter (systronics Model No-304), Optical density (O.D.) was measured on colorimeter (systronics Model No-202), Analytical balance (Wensar Model No-PGB200) was used to weigh samples and reagents, Flame photometer (systronics Model No-128) was used for analysis of Potash, Micro Nutrients was analyzed on Double beam atomic absorption spectrophotometer (Elico Model No-SL 194).

METHOD OF ANALYSIS:-

(1) Calcium

5 g air dried soil sample was taken in 150 ml conical flask and 25 ml of neutral normal ammonium acetate was added. Shaken it on mechanical shaker for 5 min, and filtered through Whatman filter paper No.1. 10 ml filtrate solution was taken in conical flask, and 2-3 crystals of sodium diethyl dithiocarbamate were added. Then 5 ml 16% sodium hydroxide and 40-50 mg of the murexide indicator were added. Titrate it with 0.01N EDTA solution till the color gradually changes from orange red to reddish violet (purple), note the titrated EDTA solution.

(2) Carbon

Method for making standard graph for Organic carbon.

Weighed out 1.25 g sucrose and taken it into 250 ml of volumetric flask and dissolved in 1 N of potassium dichromate solution, and makes up 250 ml volume by using 1 N potassium dichromate. 7 glass beakers of 50 ml were taken and numbered from 1 to 7. 0 ml, 1 ml, 2 ml, 3 ml, 4 ml, 5 ml and 6 ml solution was taken into above beakers from prepared solution of potassium dichromate. Taken 10 ml 1 N potassium dichromate solution and 20 ml conc. sulphuric acid in test-tube and placed for 30 minutes. Allowed to cool and added 20 ml distilled water. Prepared following different standard carbon ppm solution and measured optical density (O.D.) by using red filter.

Table 1: Reading for Standard Graph of Carbon

Sr.No.	ml of sucrose solution diluted in potassium dichromate	Amount of sucrose	O.D.
1	0 (blank)	-----	0
2	1	0.005 g	25
3	2	0.010 g	65
4	3	0.015 g	92
5	4	0.020 g	122
6	5	0.025 g	155
7	6	0.030 g	180
	Total	0.105 g	639

Calculation:-

1 Reading

$$\begin{aligned} 1 \text{ Reading} &= \text{Total Amount of Sucrose} / \text{Total Reading} \\ &= 0.000164319 \\ &= 0.000161043 \text{ g Sucrose} \end{aligned}$$

1 Reading Carbon value:

$$\begin{aligned} &0.00006901 \\ &0.00006764 \text{ gram organic carbon} \end{aligned}$$

$$\begin{aligned} 1 \text{ Reading Graph Factor Value} &= 0.000067638 \times 100 \\ &= 0.0067638 \end{aligned}$$

Process:

Taken 1.0 g soil sample in 100 ml beaker. 10 ml 1 N Potassium dichromate solution and 20 ml conc. Sulfuric acid were added to the sample and cooled the solution for 30 minutes. 20 ml distilled water was slowly added and allowed for 12 hrs for oxidation. Then first set zero optical density using blank solution (as above method without taking soil sample). Measured optical density (O.D.) of soil sample by using red filter and note down the reading.

(3) Sulphur

Method for making standard graph for Sulphur

Weighted out 5.434 g potassium sulphate and make up 1 Ltr by using distilled water (this solution contains 1000 ppm of sulphur). 25 ml this solution was taken and make up 1 Ltr with distilled water (this is working standard solution of sulphur). Taken 0.0 (Blank), 1.0, 2.0, 3.0, 4.0, 5.0, 6.0, 7.0, 8.0, and 10 ml working solution in 25 ml volumetric flask. In every flask 1.0 g barium chloride and 1 ml gum acacia solution were added, and make up 25 ml by using distilled water. Then optical density of blank solution was set to zero using blue filter.

Table 2: Reading for Standard Graph of Sulphur

Sr. No.	Working standard sulphur solution in ml	ppm	O.D
1	0	0	0
2	1	1	11
3	2	2	23
4	3	3	32
5	4	4	42
6	5	5	53
7	6	6	62
8	7	7	86
9	8	8	105
10	10	10	130
	Total ppm	46	Total: 544

Calculation:-

1 Reading = Total ppm of Sulphur/Total reading

$$1 \text{ Reading} = 46/544 \\ = 0.08$$

Sulphur ppm or mg/kg

Sulphur ppm or mg/kg = sample reading X graph Factor X 50 X 25 /20 X 10

Sample Reading X 0.084871 X 50 X 25/200

Sulphur ppm = Sample Reading X 0.530443 or mg/kg

Process:

10 g air dried soil sample was taken in 150 ml conical flask. 50 ml 0.15% calcium chloride extracting solution was added and shaken on mechanical shaker for 30 min. Filtered it on whatman filter No. 42. 20 ml filtrate was taken in 25 ml volumetric flask. 2 ml glacial acetic acid, 1 g crystal of barium chloride and 1 ml gum acacia solution were added. Make up the volume to 25 ml, then first set zero optical density using blank solution (as above method without taking soil sample). Measured optical densities (O.D) of above prepared sample by using blue filter.

(4) Magnesium

5 g air dried soil sample was taken in conical flask. To this, 25 ml of neutral ammonium acetate solution was added. The solution was shaken on mechanical shaker and filtered through Whatman (No.1) filter paper. 5 ml solution was pipetted out in conical flask. To this solution, 2-3 crystal sodium diethyl dithiocarbamate, 5 ml of ammonium chloride-ammonium hydroxide buffer solution and 3-4 drops of Eriochrome black-T indicator were added. Titrated it slowly against 0.01 M EDTA solution. At the end point color changed from wine red to blue.

(5) Micronutrients (Cu, Fe, Mn, Zn) analysis by AAS

Preparation of D.T.P.A extracting solution

1.967 g D.T.P.A. and 13.3 ml triethanol amine were taken in 500 ml flask. 400 ml distilled water was added. 1.47 g calcium chloride dihydrate was taken in 1ltr flask and dissolved in 400 ml distilled water. To this solution, previously prepared D.T.P.A. & T.E.A. solution was added and pH was adjusted to 7.3 by using add 1M HCl. Make up 1 ltr with distilled water.

Analysis method for micronutrients (Cu, Fe, Mn, Zn)

Weighted 20 g dried soil sample in a plastic bottle, then added 40 ml of D.P.T.A. solution. Shake on mechanical shaker for 2 hrs. Filtered it on whatman filter No. 40 in funnel cum test tube. Prepared standard curve for element by using different working ppm solution as per standard method of analysis and condition suggested by Elico brochure and then run the sample and note the ppm of elements. Obtained ppm reading multiplied with factor 2.0.

(6) Electrical Conductivity (E.C.)

10 g soil and 20 ml distilled water were taken in 50 ml beaker. It was stirred for 30 minutes. The temperature of E.C. meter was adjusted at 25 °C then conductance was adjusted to 1.412 mS/cm by using 0.01 N KCl solution. Washed the electrode with distilled water and cleaned with filter paper. Immersed electrode in above suspension solution and note the reading.

(7) pH

10 g soil & 20 ml distilled water were taken in 50 ml beaker & stirred for 30 min. In 50 ml beaker taken 10 g soil and added 20 ml distilled water and stir for 30 min. Adjusted the temperature of pH meter at 25 °C. Calibrated the pH meter using 4, 7.0, 9.2 pH buffer solution. Washed the electrode with distilled water and clean by filter paper. Immersed electrode in above suspension solution and note the reading.

(8) Phosphorus

Method for making standard graph for phosphorus.

0.439 g previously dried potassium dihydrogen orthophosphate was dissolved in 500 ml distilled water and 25 ml 7.0 N Sulphuric acid solution was added and then makes up 1 Ltr by using distilled water. 10 ml above solution was taken and makes up 500 ml by using distilled water (1 ml this resulting solution is equivalent to 2 ppm of phosphorus). By using this solution, various standard phosphorus ppm solutions were prepared and measured and their optical densities (O.D) were measured by using red filter.

Table 3 : Reading for Standard Graph of Phosphorus

Flask No	2 ppm Working Solution of Phosphorous	8.5 pH Solution of Sodium Bicarbonate	1.5 Percentage Solution of Ammonium Molybdate-HCl	Working Solution of Steanus Chloride	O.D.
1	0 Blank	5 ml	5 ml	1 ml	0
2	1 ml = 2 ppm	5 ml	5 ml	1 ml	22
3	2 ml = 4 ppm	5 ml	5 ml	1 ml	32
4	3 ml = 6 ppm	5 ml	5 ml	1 ml	52
5	4 ml = 8 ppm	5 ml	5 ml	1 ml	82
6	5 ml = 10 ppm	5 ml	5 ml	1 ml	102
7	10 ml = 20 ppm	5 ml	5 ml	1 ml	198
	Total = 50 ppm				488

Calculation

1 Reading

$$= \text{Total Solution of ppm} / \text{Total Reading}$$

$$= 50 / 488$$

$$= 0.102$$

$$0.1010 \text{ Microgram P (Graph Factor)}$$

1 Gram Soil = R X 0.1010 X 4 Microgram P/ Gram Soil

R =Colorimeter Reading of Sample 0. 1010= Graph Factor

P Kg/ Hectare = R X 0.1010 X 4 x 2.24 (2.24 = Factor in 'P' Hectare)

P₂O₅ Kg/ Hectare = R X 0.1010 X 4 x 2.24 X 2.29 (2.29 = Factor in 'P₂O₅ ' Hectare)

P₂O₅ Kg/ Hectare = R X 2.0723584

Process:

2 g soil sample and 40 ml 0.5 M sodium bicarbonate (8.5 pH) solution were taken in 100 ml beaker. To this, 1 g phosphate free activated charcoal was added and shaken on shaker for 30 minutes. The solution was filtered and pipette out 5 ml. 5ml 1.5% ammonium molybdate-hydrochloric acid solution was added to this solution. Allow to stand for 30 minutes, then 1ml 0.016 M stannous chloride solution was added & make up 25 ml using distilled water. Blank solution was prepared according to the above process without taking the soil sample. Red filter was used and zero optical density was set by using above blank solution, then put the above sample solution and note the optical density.

(9) Potassium

Method for graph factor of Potassium

Prepared following stock solution and from it make various potash ppm solutions and run in flame photometer and note down potash ppm the reading.

Table 4: Reading for Standard Graph of Potassium

Flask No.	Stock solution	Concentration of Potash in 100 ml Volumetric Solution (ppm)	Reading of Flame Photometer
1	0.0ml (Blank)	-----	0
2	1.0ml	10ppm	40
3	1.5ml	15ppm	45.5
4	2.0ml	20ppm	56.5
5	2.5ml	25ppm	62.5
6	3.0ml	30ppm	75
7	4.0ml	40ppm	100
	Total	140ppm	379.5

Calculation

$$\begin{aligned} 1 \text{ Reading} &= \text{Total Solution of ppm} / \text{Total Reading} \\ &= 140 / 379.5 \\ &= 0.369 \end{aligned}$$

$$\begin{aligned} 1 \text{ Gram Soil} &= R \times 0.369 \times 5 \text{ Microgram K} / \text{Gram Soil} \quad (0.369 \text{ Graph Factor}) \\ R &= \text{Flame Photometer Reading of sample} \end{aligned}$$

K

$$\text{Kg/Hectare} = R \times 0.369 \times 5 \times 2.24 \quad (2.24 = \text{Factor in K Hectare})$$

K₂O

$$\begin{aligned} \text{Kg/Hectare} &= R \times 0.369 \times 5 \times 2.24 \times 1.20 \quad (1.20 = \text{Factor in K}_2\text{O Hectare}) \\ &= R \times 4.959 \end{aligned}$$

Process:

5 g soil sample was taken in 100 ml conical flask. 25 ml 1 M neutral ammonium acetate solution was added. Shaken it for 5 minutes on shaking machine and filtered the solution on whatman filter paper. Flame photometer was calibrated by using 10, 20, 30, 40, 50, 60, 70, 80 and 90 ppm standard potassium solution. After calibration run above filtrate for analysis and note down the reading.

Result and Discussion

Soil sampling

Soil sampling was done during the dry season. Soil sampling was done at five randomly located points within each farm. The soils were sampled at two depths, 0 to 15 cm, 15 to 35 cm, using mini-soil pits dug at each sampling point. The soil samples were air dried in the laboratory and sieved through a 2 mm sieve for different types of laboratory analyses. The Results of soil samples & its LMH data shown in table no: 7(A), 7(B), 8(A), 8(B), 9(A), 9(B), 10(A), 10(B), 11(A), and 11(B).

Table 5: Critical Limits of Nutrients:-

Sr. No	Parameters	Unit	Critical Limits		
			Low	Medium	High
1	pH	-----	<6.5	6.5-8.2	>8.2
2	Electric Conductance	-----	<1	1-3	>3
3	Organic carbon	%	<0.51	0.51-0.75	>0.75
4	Phosphorous	Kg/Hectare	<26	26-60	>60
5	Potash	Kg/Hectare	<151	151-300	>300
6	Zinc	ppm	<0.5	0.5-1.0	>1.0
7	Ferrous	ppm	<5	5-10	>10
8	Sulphur	ppm	<10	10-20	>20
9	Manganese	ppm	<5	5-10	>10
10	Copper	ppm	<0.2	0.2-0.4	>0.4
11	Magnesium	ppm	<1.0	1.0-2.0	>2.0
12	Calcium	ppm	<1.5	1.5-3.0	>3.0

Calculation of soil fertility Index:

$$= \frac{(\% \text{ of Low} \times 1) + (\% \text{ of Medium} \times 2) + (\% \text{ of High} \times 3)}{100}$$

Table No 6: Calculation of Low, Medium, and high rating of soil fertility Index:

Sr. No.	Rang	Rating
1	Less than 1.67	Low
2	1.67 to 2.33	Medium
3	Greater than 2.33	High

Table 7(A): Analysis of soil sample

Samples site: Village: Bhalgam, Taluka: Vadgam, District: Banaskantha, Gujarat, India.

Sample No.	pH	EC	Org. Carbon (%)	Phosphorous (Kg/ Hectare)	Pottash (Kg/ Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/100 g soil)	Ca (Me*/100 g soil)
1	8.76	0.24	0.41	24.87	255.67	0.38	4.94	7.96	8.42	0.98	3.05	7.25
2	8.40	0.57	0.28	26.94	704.10	0.52	3.88	10.61	16.02	0.98	2	6.6
3	8.37	0.33	0.46	29.01	266.38	0.28	4.44	16.44	8.82	0.6	2.05	4.35
4	8.46	0.65	0.30	31.09	649.22	0.54	4.8	26.52	15.74	0.82	2.55	5.85
5	8.72	0.24	0.26	22.80	215.51	0.46	5.62	24.40	7.84	0.54	2.25	5.05
6	8.45	0.59	0.23	51.81	625.13	0.54	4.38	146.93	12.22	0.96	1.75	5.65
7	8.80	0.24	0.32	41.45	274.41	0.52	6.6	1.06	12.54	1.06	2.3	6.4
8	8.45	0.55	0.35	43.52	724.18	0.5	7.72	116.17	19.76	1.38	1.4	5.8
9	8.75	0.24	0.28	45.59	254.33	0.26	3.82	18.57	9.46	0.72	2.2	6.1
10	8.81	0.25	0.83	31.09	318.59	0.24	4.16	23.34	10.84	0.7	2.3	5.8
11	8.49	0.63	0.57	35.23	271.74	0.68	4.3	123.06	9.52	0.96	1.35	5.85
12	8.91	0.25	0.60	66.32	215.51	0.46	4.44	26.52	12.32	1.1	2.55	8.25
13	8.92	0.20	0.49	64.24	234.26	0.48	5.62	11.14	11.8	1.06	2.6	5.8
14	8.70	0.29	0.45	41.45	186.07	0.34	5.62	13.26	8.08	0.78	1.7	5.8
15	8.95	0.21	0.49	43.52	250.32	0.38	4.1	13.79	10.84	0.88	2.05	6.05
16	9.00	0.25	0.64	47.66	216.85	0.28	4.88	23.87	7.66	0.82	2.05	8.25
17	8.68	0.25	0.53	49.74	285.12	0.46	5.54	16.44	18.84	0.82	2.55	11.05
18	8.77	0.31	0.58	31.09	234.26	0.26	4.38	17.50	6.76	0.7	1.95	9.55
19	8.84	0.24	0.40	29.01	218.19	0.32	4.24	20.16	4.42	0.78	2.85	5.65
20	8.78	0.24	0.35	26.94	211.50	0.6	4.1	13.79	8.62	0.54	0.95	6.15

*=Miliequivalent

Table 7(B): Soil Fertility Index & Soil Test Rating

Samples site: Village: Bhalgam, Taluka: Vadgam, District: Banaskantha, Gujarat, India.

Sample No.	pH	EC	Org. Carbon (%)	Phosphorous (Kg/Hectare)	Pottash (Kg/Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/100 g soil)	Ca (Me*/100 g soil)
L	0	20	14	2	0	13	14	2	1	0	1	0
M	0	0	5	16	15	7	6	9	9	0	6	0
H	20	0	1	2	5	0	0	9	10	20	13	20
%L	0	100	70	10	0	65	70	10	5	0	5	0
%M	0	0	25	80	75	35	30	45	45	0	30	0
%H	100	0	5	10	25	0	0	45	50	100	65	100
S.F.I.*	3.00	1.00	1.35	2.00	2.25	1.35	1.30	2.35	2.45	3.00	2.60	3.00
LMH** of SFI	H	L	L	M	M	L	L	H	H	H	H	H

*= Soil Fe ppm Fertility Index, **= Low, Medium, and High Soil Fe ppm Fertility Index

CONCLUSION:- Above results indicate that E.C., Org. Carbon, Zn & Fe are in low amount, Farm yard manure & Zinc Sulphate and Ferrous Sulphate should be added for better plant growth & productivity. The other parameters are sufficient. pH is in high limit. So it can be neutralized by using acidic fertilizer.

Table 8(A): Analysis of soil sample

Samples site: Village: Nagarpura, Taluka: Vadgam, District: Banaskantha, Gujarat, India.

Sample No.	pH	EC	Org. Carbon (%)	Phosphorous (Kg/ Hectare)	Pottash (Kg/Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me/100 g soil)	Ca (Me/100 g soil)
1	7.85	0.18	0.38	43.52	165.99	0.98	4.9	19.10	17.04	0.8	2.15	5.55
2	7.75	0.17	0.45	49.74	240.95	1.02	5.32	21.22	17.76	0.92	1.85	7.05
3	8.15	0.24	0.70	31.09	196.77	2.54	3.96	28.11	20.08	0.82	3.35	7.15
4	8.25	0.17	0.65	22.80	323.94	2.28	5.32	33.42	17.62	0.86	2.65	6.85
5	7.96	0.17	0.41	29.01	182.05	1.04	5.12	18.04	16.5	0.72	3.85	8.65
6	8.50	0.14	0.39	26.94	210.16	1.08	4.64	12.73	18.24	0.84	1.1	7.1
7	8.00	0.16	0.49	22.80	144.57	0.86	4.56	19.10	15.5	0.82	1.25	6.55
8	8.27	0.24	0.66	31.09	311.89	1.3	2.92	14.32	13.78	0.76	0.85	6.65
9	8.25	0.15	0.30	26.94	307.88	1.02	3.7	14.32	11.86	0.56	1.85	8.65
10	8.45	0.19	0.52	64.24	192.76	2.34	6.52	14.85	26.16	1.08	2.75	8.05
11	8.30	0.15	0.45	49.74	270.40	1.8	6.52	24.93	17.32	0.94	2.7	8.4
12	8.00	0.16	0.41	41.45	203.47	1.48	9.58	16.97	19.58	1.08	3.15	9.05
13	8.20	0.14	0.64	26.94	191.42	1.32	7.02	29.17	17.52	0.92	3.15	7.15
14	7.90	0.19	0.40	24.87	207.48	1.1	7.24	38.72	16.16	0.96	2.55	6.95
15	7.95	0.18	0.43	43.52	195.44	1	7.24	19.63	17.32	1.04	2.5	6.5
16	7.86	0.20	0.49	31.09	183.39	1.22	6.66	18.04	17.96	1.04	1.5	6.6
17	7.93	0.14	0.34	33.16	144.57	1.02	5.32	21.22	17.32	1	0.95	6.75
18	7.90	0.16	0.36	64.24	155.28	0.98	5.94	19.63	14.84	0.92	3.05	6.05
19	7.96	0.16	0.30	82.89	188.74	1.06	6.8	20.69	19.44	0.98	3.65	5.95
20	8.00	0.15	0.51	70.46	220.87	1.14	6.58	22.81	18.2	1	2.15	6.95

*=Miliequivalent

Table 8(B): Soil Fertility Index & Soil Test Rating

Samples site: Village: Nagarpura, Taluka: Vadgam, District: Banaskantha, Gujarat, India.

Sample No.	pH	EC	Org. Carbon (%)	Phosphorous (Kg/Hectare)	Pottash (Kg/Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me/100 g soil)	Ca (Me/100 g soil)
M	14	0	5	13	15	4	14	11	0	0	5	0
H	6	0	0	4	3	16	0	9	20	20	13	20
%L	0	100	75	15	10	0	30	0	0	0	10	0
%M	70	0	25	65	75	20	70	55	0	0	25	0
%H	30	0	0	20	15	80	0	45	100	100	65	100
S.F.I.*	2.3 0	1.00	1.25	2.05	2.05	2.80	1.70	2.45	3.00	3.00	2.55	3.00
LMH** of SFI	M	L	L	M	M	H	M	H	H	H	H	H

*= Soil Fertility Index, **= Low, Medium, and High Soil Fertility Index

CONCLUSION:- Above results indicate that E.C. and Org. Carbon are in low amount, Farm yard manure should be added for better plant growth & productivity. The other parameters are sufficient. pH is in medium limit, So it can be neutralized by using acidic fertilizer.

Table 9(A): Analysis of soil sample

Samples site: Village: Shamsherpura, Taluka: Vadgam, District: Banaskantha, Gujarat, India.

Sample No.	pH	EC	Org. Carbon (%)	Phosphorous (Kg/Hectare)	Pottash (Kg/Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/100 g soil)	Ca (Me*/100 g soil)
1	8.35	0.23	0.57	43.52	373.47	0.96	6.24	0.53	15.36	0.92	1.8	4.2
2	8.45	0.17	0.48	49.74	206.14	0.62	5.32	26.52	12.46	0.66	1.8	4.9
3	8.41	0.18	0.43	29.01	298.51	0.62	5.04	23.87	12.2	0.8	2.2	6.4
4	8.61	0.19	0.46	20.72	357.41	0.54	4.62	28.64	16.08	0.64	2.05	5.05
5	8.30	0.20	0.51	31.09	376.15	0.94	4.82	21.75	15.02	0.66	1.6	4.9
6	9.00	0.21	0.55	43.52	259.69	1.12	6.52	26.52	19.2	0.8	1.45	4.35
7	8.40	0.18	0.37	51.81	240.95	0.58	5.04	43.50	14.04	0.64	2.25	4.95
8	8.30	0.26	0.40	49.74	321.26	0.8	6.24	25.46	14.74	0.78	2.25	6.95
9	8.35	0.25	0.72	43.52	358.74	0.98	5.32	21.22	15.76	0.44	1.45	3.25
10	9.20	0.58	0.62	45.59	270.40	1.94	6.68	17.50	17.5	0.54	2.3	3.8
11	8.32	0.21	0.59	41.45	405.60	1.18	5.46	14.85	15.24	0.66	1.1	4.8
12	8.50	0.18	0.47	31.09	262.37	0.84	5.24	18.04	11.82	0.52	0.55	7.55
13	8.57	0.18	0.30	47.66	216.85	0.92	6.82	29.70	14.88	0.52	1.95	3.35
14	8.40	0.21	0.45	29.01	396.23	0.74	6.82	16.44	19.72	0.72	1.85	6.05
15	8.50	0.24	0.46	64.24	315.91	1.52	6.6	18.57	19.2	0.72	2.1	5.7
16	8.28	0.25	1.00	49.74	414.97	1.78	6.9	33.42	21.84	0.8	2.75	6.55
17	8.32	0.20	0.74	43.52	295.83	0.76	6.52	19.10	14.44	0.7	2.1	8.9
18	8.36	0.18	0.31	68.39	208.82	0.94	7.34	21.22	18.5	0.9	4.35	10.35
19	8.26	0.43	1.31	31.09	421.66	2.16	7.56	24.40	23.6	0.38	3.35	8.25
20	8.00	0.66	0.93	35.23	404.26	1.66	6.68	33.42	24.14	0.58	3.1	8.8

***=Miliequivalent**

Table 9(B): Soil Fertility Index & Soil Test Rating

Samples site: Village: Shamsherpura, Taluka: Vadgam, District: Banaskantha, Gujarat, India.

Sample No.	pH	EC	Org. Carbon (%)	Phosphorous (Kg/Hectare)	Pottash (Kg/Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/100 g soil)	Ca (Me*/100 g soil)
L	0	20	10	1	0	0	2	1	0	0	1	0
M	1	0	7	17	9	13	18	6	0	1	8	0
H	19	0	3	2	11	7	0	13	20	19	11	20
%L	0	100	50	5	0	0	10	5	0	0	5	0
%M	5	0	35	85	45	65	90	30	0	5	40	0
%H	95	0	15	10	55	35	0	65	100	95	55	100
S.F.I.*	2.95	1.00	1.65	2.05	2.55	2.35	1.90	2.60	3.00	2.95	2.50	3.00
LMH** of SFI	H	L	L	M	H	H	M	H	H	H	H	H

*= Soil Fertility Index, **= Low, Medium, and High Soil Fertility Index

CONCLUSION:- Above results indicate that E.C. and Org. Carbon are in low amount, farm yard manure should be added for better plant growth & productivity. The other parameters are sufficient. pH is in high limit, So it can be neutralized by using acidic fertilizer.

Table 10(A): Analysis of soil sample

Samples site: Village: Jalotra, Taluka: Vadgam, District: Banaskantha, Gujarat, India.

Sample No.	pH	EC	Org. Carbon (%)	Phosphorous (Kg/Hectare)	Pottash (Kg/Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/100 g soil)	Ca (Me*/100 g soil)
1	8.49	0.17	0.42	89.11	253.00	0.4	4.6	29.17	15.44	1.1	2.3	5.7
2	8.64	0.13	0.56	64.24	153.94	0.34	4.24	39.78	10.48	1.04	2.05	6.55
3	8.35	0.13	0.64	82.89	182.05	0.3	6.04	23.34	18.52	1.38	1.8	6.3
4	8.44	0.12	0.60	87.04	164.65	0.26	5.24	33.95	15.08	1.18	2.3	6.9
5	8.27	0.11	0.63	89.11	188.74	0.28	5.08	29.70	19.54	1.1	2.95	6.75
6	8.27	0.10	0.74	82.89	176.70	0.2	4.2	29.17	11.66	1.08	2.95	7.15
7	8.13	0.16	0.55	87.04	242.29	0.26	3.68	25.46	10.62	0.98	1.85	5.55
8	8.42	0.12	0.52	91.18	192.76	0.36	4.3	32.36	11.98	1.04	3.05	6.85
9	8.62	0.11	0.48	82.89	145.91	0.28	4.24	42.44	12.32	1.12	4.1	7.5
10	8.64	0.13	0.43	66.32	179.37	0.3	4.04	35.54	11.54	1.04	4.65	8.65
11	8.61	0.14	0.47	45.59	143.23	0.24	4.2	53.04	10.96	1.1	3	5.5
12	8.53	0.13	0.46	82.89	179.37	0.24	3.84	30.24	13.1	1.12	1.4	4.8
13	8.54	0.12	0.41	87.04	170.00	0.24	4.7	31.83	12.9	1.08	1.75	5.15
14	8.35	0.21	0.51	49.74	159.29	0.26	3.74	29.70	9.86	1.06	1.25	5.85
15	8.45	0.11	0.49	84.97	149.92	0.24	4.24	40.84	12.74	1.1	1.9	8.2
16	8.70	0.16	0.55	70.46	191.42	0.28	4.4	44.56	10.08	1.16	1.9	9.2
17	8.80	0.12	0.51	87.04	160.63	0.2	3.78	23.87	9.58	1	2.45	5.05
18	8.25	0.14	0.66	62.17	212.84	0.3	5.18	28.64	14.26	1.22	2.2	6.3
19	8.60	0.12	0.50	49.74	170.00	0.2	4.66	37.66	13.48	1.12	2.45	6.65
20	8.60	0.10	0.20	29.01	101.73	0.16	3.1	38.72	5.88	0.72	2.55	5.15

*=Miliequivalent

Table 10(B): Soil Fertility Index & Soil Test Rating

Samples site: Village: Jalotra, Taluka: Vadgam, District: Banaskantha, Gujarat, India.

Sample No.	pH	EC	Org. Carbon (%)	Phosphorous (Kg/Hectare)	Pottash (Kg/Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/100 g soil)	Ca (Me*/100 g soil)
L	0	20	10	0	4	20	16	0	0	0	0	0
M	1	0	10	4	16	0	4	0	3	0	7	0
H	19	0	0	16	0	0	0	20	17	20	13	20
%L	0	100	50	0	20	100	80	0	0	0	0	0
%M	5	0	50	20	80	0	20	0	15	0	35	0
%H	95	0	0	80	0	0	0	100	85	100	65	100
S.F.I.*	2.95	1.00	1.50	2.80	1.80	1.00	1.20	3.00	2.85	3.00	2.65	3.00
LMH** of SFI	H	L	L	H	M	L	L	H	H	H	H	H

*= Soil Fertility Index, **= Low, Medium, and High Soil Fertility Index

CONCLUSION:- Above results indicate that E.C., Org. Carbon, Zn and Fe are in low amount, farm yard manure, Zinc Sulphate and Ferrous Sulphate should be added for better plant growth & productivity. The other parameters are sufficient. pH is in high limit, So it can be neutralized by using acidic fertilizer.

Table 11(A): Analysis of soil sample

Samples site: Village: Pepol, Taluka: Vadgam, District: Banaskantha, Gujarat, India.

Sample No.	pH	EC	Org. Carbon (%)	Phosphorous (Kg/Hectare)	Pottash (Kg/Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/100 g soil)	Ca (Me*/100 g soil)
1	8.40	0.61	0.65	82.89	441.74	0.68	5.46	15.38	17.56	0.92	2.65	4.95
2	8.11	0.33	0.28	43.52	137.88	0.5	5.98	112.45	9.66	0.82	2.85	6.05
3	8.22	0.31	0.66	87.04	144.57	0.42	6.52	16.97	7.98	0.92	3.7	6.9
4	8.14	0.24	0.47	26.94	152.60	0.54	5.02	7.43	6.7	0.98	2.45	6.25
5	7.97	0.85	0.60	41.45	496.62	0.64	6.14	6.90	16.26	1.02	3.35	6.55
6	7.97	0.80	4.83	66.32	155.28	0.4	5.32	126.25	7.22	0.78	2.55	5.85
7	8.15	0.37	0.45	26.94	543.47	0.7	7.16	20.16	19.58	0.92	3.3	5.3
8	8.00	0.24	0.49	29.01	222.21	0.48	5.32	138.45	9.52	0.76	3.9	5.7
9	8.60	0.22	0.74	41.45	216.85	0.4	4.58	23.34	10.38	0.82	4.8	5.6
10	8.11	0.22	0.43	72.53	479.22	0.56	5.84	17.50	12.6	1.06	1.85	5.45
11	7.96	0.78	0.58	26.94	526.07	0.76	4.44	14.32	20.14	1.28	0.7	6.8
12	8.29	0.22	0.45	41.45	161.97	0.4	5.32	137.92	7.5	0.92	0.35	4.65
13	8.12	0.24	0.64	62.17	244.96	0.42	5.32	117.76	12.32	0.96	1	5.5
14	8.22	0.22	0.51	26.94	147.25	1.1	8.04	92.83	12.44	1.54	0.35	5.15
15	8.21	0.32	0.68	29.01	208.82	0.8	8.46	18.57	17.92	1.28	0.55	6.75
16	8.90	0.33	0.49	64.24	141.89	0.66	9.25	10.61	11.64	0.92	0.95	5.85
17	8.16	0.26	0.66	62.17	211.50	0.74	8.54	8.49	14.88	1.32	1.4	5.2
18	8.00	0.78	0.54	45.59	473.86	0.92	8.34	11.14	22.42	1.36	3.25	5.25
19	8.22	0.27	0.69	47.66	236.93	0.6	8.88	5.30	13.04	1.06	3.45	5.25
20	8.32	0.24	0.57	29.01	198.11	0.64	9.92	16.97	14.7	1.1	5.3	3.5

*=Miliequivalent

Table 11(B) : Soil Fertility Index & Soil Test Rating

Samples site: Village: Pepol, Taluka: Vadgam, District: Banaskantha, Gujarat, India.

Sample No.	pH	EC	Org. Carbon (%)	Phosphorous (Kg/Hectare)	Pottash (Kg/Hectare)	Zn ppm	Fe ppm	Sulphur ppm	Mn ppm	Cu ppm	Mg (Me*/100g soil)	Ca (Me*/100g soil)
L	0	20	7	0	4	6	2	4	0	0	5	0
M	11	0	12	13	10	13	16	8	6	0	3	0
H	9	0	1	7	6	1	2	8	14	20	12	20
%L	0	100	35	0	20	30	10	20	0	0	25	0
%M	55	0	60	65	50	65	80	40	30	0	15	0
%H	45	0	5	35	30	5	10	40	70	100	60	100
S.F.I.*	2.45	1.00	1.70	2.35	2.10	1.75	2.00	2.20	2.70	3.00	2.35	3.00
LMH** of SFI	H	L	M	H	M	M	M	M	H	H	H	H

*= Soil Fertility Index, **= Low, Medium, and High Soil Fertility Index

CONCLUSION:- Above results indicate that E.C. is in low amount. The other parameters are sufficient. pH is in high limit, So it can be neutralized by using acidic fertilizer.

Reference:

1. S. P. Friedman, *Computers and Electronics in Agriculture*, 46 45–70, 2005.
2. T. J. Kelleners, D. A. Robinson, P. J. Shouse, J. E. Ayars and T. H. Skaggs, *Soil Sci. Soc. Am. J.*, 6967–76,2005.
3. T. Saarenketo , *Journal of Applied Geophysics*, 4073–88, 1998.
4. J.M. Sabatier, D.C. Sokol, C.K. Frederickson, M.J.M. Rdmkensb, E.H. Grissingerb, J.C. Shipps, *Soil Technology* ,8259-274, 1996.
5. J. Yick, B. Mukherjee and D. Ghosal, *Computer Networks*, 52 2292–2330, 2008.
6. Verstraeten, W.W.; Veroustraete, F.; Wagner, W.; Van Roey, T.; Heyns, W.; Verbeiren, S.; Van der Sande, C.J.; Feyen, J.; IIASA, Austria, 27-28 Sep, 2007.
7. P. G. Adsule, *J food science techno*, vol. 16, , pp. 216 – 217, 1979.
8. Epstein W. Sunderland MA: Sinauer Ass. Inc:, 2005.
9. Beaufils ER. *Soil Science Bulletin*, 1, Pietermaritzburg: University of Natal;,1973.